Aquatic Rehabilitation for Medically Fragile and Terminally Ill Children: A Case Study

Effects of Water Exercise on Muscle Strength and Endurance

Aquatic Rehabilitation for Orthopedic Trauma: Part One
Genetic Disease Center. Out of what she learned from Sophia’s struggle for survival and her family's search for meaning in the midst of uncertainty, she created Healing in Community, a compassionate and highly effective community-based approach to caring for children with life-threatening conditions and their families. She co-founded Sophia's Garden Foundation to share this knowledge with the world.

**Reviewer Comments**

**Julie Meno Fettig**

This article brings awareness of the role aquatic therapy can play in management of pain and providing palliative care. Sophia's case study is an excellent example of how coordinated professional disciplines, different services, and timing of treatment need to be for successful aquatic therapy with a medically fragile child. Warm water aquatic therapy is an excellent pain management modality. Aquatic therapy for the treatment of pain has greatest benefit when water is clean, temperature correct for patients condition, and environment is calm. The therapist should be knowledgeable about the specific condition, adaptable, reassuring, empathetic, yet humorous. This three dimensional supporting environment can be a great equalizer against pain. When suspended in water, without fear, it allows us to feel and sense ourselves from within. From within we can heal ourselves, feel strong, in control, and very much alive.

**Reviewer Bios**

**Julie Meno Fettig, CTRS, ATRIC**, is the founder/owner of Therapeutic Aquatics, Inc. and aquaticcentral.com, specializing in consulting, information, and rehabilitation. She is the author and publisher of The Bad Ragaz Ring Method Visual Instructional Manual and video and co-producer of the PNF in the Pool video. She received the 2002 ATRI Tsunami Spirit Award.

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**Feature Column: Pool Problems**

**Cloudy Pool Water**

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Aquatic professionals often notice that pool water becomes turbid gradually throughout the day or immediately after lengthy periods of peak use. Cloudy water conditions may also occur immediately after chemical adjustments are made. Water may appear cloudy or milky. A fine white precipitate may settle out of the water. Water clarity frequently deteriorates to the point where it is not safe to continue operation and classes or programs must be cancelled.

Although water clarity should exceed 0.25 NTUs (Nephelometric Turbidity Units), most public pool bathing codes permit swimming pools to be used until the clarity deteriorates to the point that either the main drains or a 6-inch diameter black disk can't be clearly seen from the surrounding deck at the deepest point in the pool, or the black and red (or black and white) quarter panels on a 2-inch diameter, Secchi disk cannot be distinguished at a depth of 15 feet.

Aquatic professionals should insist water clarity be maintained within an acceptable range. It should be understood that activities will be cancelled rather than endangering users. Written pool rental agreements should outline how the instructor or therapists will be compensated for lost fees or wages if the pool is not able to be used. If cloudy water problems result in more than very infrequent pool closures, the cause of the problem should be identified and remedied.

Water clarity problems can usually be traced to one of two possible causes – either physical or chemical in nature. Physical problems are caused by the design of the circulation system or mis sized equipment. Chemical problems usually result from improper application of chemicals, incorrect dosing, or from not correcting water quality problems when they occur.

**Chemical Problems**

Sometimes chemicals are added to water in too great a quantity in too short a period of time. With the exception of chlorine, pool chemicals should be added to the pool gradually, and in small quantities over an extended period of time. Pre dissolve solid, granulated or powdered chemicals prior to their addition. Try to limit chemical additions to 10 ppm changes at a time.

**Excessively high Total Dissolved Solids (TDS)** can cause water to appear less than crystal clear. Use a TDS meter to determine the level of total dissolved solids. In pools with high bather load to water volume ratios, regular dilution is recommended at a rate of 8 gallons per pool user per day. If TDS levels exceed 1,500 ppm and are causing problems with taste, clarity, ability to maintain ORP levels, or galvanic corrosion, dilute significantly, or drain and refill the pool with fresh water.

High concentrations of cyanuric acid will interfere with oxidation of organic contaminants in the water. Do not use cyanuric acid or chlorinated isocyanurates, such as trichloro-s-triazinetrione or sodium dichloro-s-triazinetrione, in indoor pools, or in outdoor pools and spas with extremely high organic loading problems. If cyanurates are used to prevent loss of chlorine and dissipation into the air due to exposure to ultraviolet light, use them in moderation. Keep cyanuric acid levels in the 10 ppm – 20 ppm range since 95% of the staying power benefit is achieved in that range. Also, the negative effects on pathogenic organism kill time and depression of ORP are still within an acceptable range.
If water is difficult to balance due to extremely high calcium hardness levels in the source water, use of sequestering or chelating agents is recommended. Sequestering agents increase the ability or chelating agents is recommended. If water is difficult to balance due to extremely high calcium hardness levels when pH and water temperature rise, discoloring or clouding the water, attaching to and discoloring bathers’ hair. Chelating agents remove metals or dissolved minerals from the water. They cure mineral staining problems. Organic water soluble molecules bond and react with ions to keep them from precipitating.

**Oversaturated water** is one of the most frequent causes of cloudy pool water. Since water is the universal solvent, all things will inevitably dissolve in water until the water becomes saturated. Eventually, water will become unbalanced or oversaturated and excess products will be lost by precipitation. Well balanced water will increase bather comfort and will dramatically extend the life expectancy of the pool and its components. Water temperature, pH, total alkalinity, calcium hardness, and total dissolved solids act together to cause corrosiveness or calcification qualities of water. The Langelier Saturation Index formula and chart can be used to determine if pool water is balanced — that is, neither aggressive nor oversaturated.

To calculate the saturation index, use the formula: Saturation index equals pH plus the alkalinity factor, plus the calcium hardness factor, plus the temperature factor, minus the TDS factor.

Use your test kit and testing instruments to find each of the five values. Write down the actual pH value found. Then for the remaining four values, find the corresponding factor on the chart. Add or subtract the factors to or from the pH value. If an actual value is not found on the saturation index chart, do not interpolate since there is no direct linear relationship between the values. Rather, move to the next higher value and use its factor.

If the sum obtained is zero, the water is balanced and chemical equilibrium has been achieved. A tolerance of plus or minus 0.3 is allowable for commercial swimming pools. Negative values indicate aggressive water, while positive values indicate likely calcification and scale formation.

Undersaturated water is aggressive and will cause circulation pipes, heater elements, and other metal components of the pool to corrode. Pool wall surface materials will deteriorate. Plaster will soften and etch, vinyl liners will become brittle, metal staining will increase, and tiles will become loose and begin popping off the walls.

If water is oversaturated, calcium carbonate will begin to settle out of the water. Water will become cloudy and take on a milky appearance. Scale will build up on solid surfaces, making surfaces rough, and discoloring dark surfaces. Calcium carbonate scale will also build up on interior surfaces of the pool recirculation pipes, restricting flow and increasing water pressure. Sanitizer effectiveness will be reduced, and algae growth may increase.

If the saturation index formula indicates the pool water is not balanced, make the appropriate chemical corrections, starting with total alkalinity, then followed by pH, temperature, calcium hardness, and TDS.

**Algae** blooms may cause pool water to become turbid, cloudy, or discolored. Algae is a waterborne plant introduced into pools by swimmers, make-up water, rain, wind and windborne debris. Although algae in and of itself is not harmful to swimmers, it does cause problems when allowed to grow in a swimming pool. Algae gradually removes carbon dioxide from the water in order to manufacture food and may cause a dramatic rise in pH. Pool surfaces can become slippery from a noticeable algae growth on the pool bottom or walls. Algae is a higher organism that may harbor pathogens or disease causing bacteria. Chlorine demand may be high, as chlorine is used in an attempt to kill or control algae growth. Pools filled with algae may give off unpleasant odors.

To control algae growth, maintain adequate chlorine and oxidation reduction potential (ORP) levels, keep the water circulating continuously, make sure you have a uniform circulation pattern and absence of dead spots in the pool, superchlorinate regularly, and scrub or brush pool walls to prevent algae from adhering. If water is not continuously circulated, sanitized and oxidized, you may need to use commercially prepared algacides or algacets to keep algae growth under control. Some algacides are more effective against a particular type of algae, and some are more appropriate for use in pools or in spas.

If you continue to have serious algae problems, you may want to monitor nitrate levels more closely, and try to determine the source of contamination. Nitrites stimulate plant growth, and when high levels of nitrates (greater than 25 ppm) are present in pool water, uncontrolled algae growth often occurs even though unaccountably large amounts of chlorine are being used.
Nitrates are introduced into pools from: fill water in areas where fertilizer has worked its way down into the ground water, contaminated reservoirs or wells, rain, fertilizers or grass blown into the pool from the adjoining landscaping, human or animal urine or fecal matter, and bird droppings. Pools located in agriculture areas, screened pools, and pools that border large bodies of water often experience nitrate problems. To lower pool nitrate levels, try shocking the pool with chlorine to over 30 ppm, or partially drain and refill the pool with water not contaminated with nitrates.

**Physical Problems**

Water clarity problems may be persistent if the pool circulation and filtration system was not properly designed or if the components were incorrectly sized. However, even the best designed system will not keep water sparkling clear if components are not properly maintained, or programming and bather loads increase beyond expectation.

To maintain clarity, keep bather load to total filtered water in gallons per day ratio at 1 bather : 1,400 gallons or less. The onset of turbidity is constant and related to the number of bathers, not just turnover time. If debris is added to the pool water faster than the filter can remove it, turbidity will increase. Debris is introduced into a pool through airborne dirt, dust, plant matter, and pollen; rain water, and bathers. But the greatest amount of debris is brought into the pool by bathers.

To determine maximum bather load: multiply flowrate (gpm) x 60 (minutes/hour) x 24 (hours/day) to get the total filtered gallons per day. Then, divide total filtered gallons per day by the constant 1,400 gallons to get the maximum number of bathers per day who can enter the pool before water clarity problems result.

To find the needed turnover time required at a given maximum bather load: Multiple the actual number of bathers using the pool per day by the constant 1,400 to get total filtered gallons per day needed. Divide by 24 (hours/day), then divide by 60 (minutes/hour) to get the required flowrate in gpm. Divide the volume of the pool in gallons by the required flowrate to get the needed turnover time in minutes.

Interestingly, since there are 1,440 minutes in a day, a short cut method of determining the correct turnover time is simply to equate bather load and flow rate. The flowrate in gallons per minute and the maximum bather load should be similar. For example, a pool with a maximum daily bather load of 300 swimmers should have a flowrate of around 300 gallons per minute in order to maintain good water clarity.

Know your pool's baseline readings, and monitor turnover time. Read the flowmeter and pressure gauges daily and record the results. Make sure that the normal flowrate is being maintained and that an obstruction or pump impeller damage due to cavitation is not restricting the amount of water moving through the filters.

Make sure filters are properly sized. If water is allowed to flow through the media at a rate higher than recommended by the manufacturer and NSF International, debris will pass right through without being removed. To determine needed filter size, calculate square footage of each filter tank (or look on the permanently affixed plate on the front of the filter). Take the flowrate in gallons per minute (gpm) and divide by the design flow rate for the particular tank using the same media. The total media square footage should exceed this number.

For example, a pool with a flowrate of 1,000 gpm, is being filtered with a bank of 4 horizontal high rate sand filters each with 13.5 square feet of #20 silica sand filter media for a total of 54 square feet of media. The design flow rate is a minimum 15 gpm/ft². One thousand gpm divided by 15 gpm/ft² equals 66.6 ft². The filters are considerably undersized and water is likely to be cloudy during periods of heavy use.

Assure that all valves are open or in the correct position to allow water to move through filters. Label all valves, and post a diagram on the pump room wall showing the correct position of valves during normal operation and during backwash procedures.

Broken laterals inside of a filter tank can allow debris to enter the pool and will cause a loss of filter media available for filtering particles from the water. Check bottom of the swimming pool first thing in the morning before the water has been agitated and look for regular deposits of filter media near the return inlets. Isolate individual filter tanks from the bank to try to determine which laterals have broken. Remove the filter media and inspect the laterals at the bottom of the tank, replacing those which have broken.

If filters are not backwashed properly and for an adequate amount of time, fine particles start to work their way down into the filter bed. Eventually fines are carried into the laterals and back into the pool. On filter systems with automatic backwash valves, make sure booster pumps are bringing the pressure up to 50 psi during the backwash process.

Perform regular filter tank inspection and maintenance on a monthly basis. Open the filter tank and make observations, being careful not to damage the filter tank or components. Dig or poke around with a trowel and look for: flatness of the media bed, channeling (holes), biofilms on the tank walls, media migration, and contamination caused by improper backwashing or improper chemical balance.

While the tank is open for inspection, perform a settling test to determine make-up of the filter bed. Take a large glass jar (like a mayonnaise jar) and fill it with 2 cups of water. Add 1 cup of media from your filter. Add 1 teaspoon of dishwasher detergent or Calgon water softener. Replace jar lid and shake. Allow the solution to settle overnight. The sample should settle into a layer of sand with water on top. If instead, it settles into layers with sand on the bottom, silty material above the sand layer, and an organic layer on top, replace the filter media in the tanks.

Clean the sand media inside the filter tank by adding a commercial sand
Mudballs and channels which form inside the sand should be destroyed. Mudball formation is caused by calcium scale, organic debris, detergents, oils, and bather waste products. These oily products reduce sanitizer effectiveness, scale, organic debris, detergents, oils, and contribute to mudball formation in sand filters causing reduced filter effectiveness.

Use of enzymes or absorbent foam products is recommended to help prevent filter problems from occurring in the first place. Enzymes are catalysts that start or speed up chemical reactions. Enzymes are protein-like substances that form naturally in animal and plant cells, but today, synthetic enzymes have been developed. Enzymes slowly, over several days, digest and destroy oils in pool water by converting them to carbon dioxide and water. A similar process is used to clean up oil spills occurring in the ocean. An initial dose of one to two ounces of enzyme per 1,000 gallons of pool water is recommended, and then maintenance doses of about half that amount should be added to the pool on a weekly basis.

Absorbent foam products can also be used to physically remove oils from the water. Manufacturers of the products say the patented molecular structure and cell design of the foam allows it to absorb many times its own weight in oil. When the foam is saturated with oil, it turns a dark color, becomes heavy and sinks. The foam can be replaced or, for a period of time, can be cleaned and reused by removing the absorbent foam from the pool skimmer, hair and lint strainer or filter tank, squeezing out the oils and replacing it in its hidden location.

Colloidal particles are particles smaller than 1 micron in size, which are suspended in water. Colloids are small enough to pass through pool filters, too light to settle on the bottom of the pool, and make water murky or cloudy. Flocculants and clarifiers make colloidal particles stick together or coagulate so that the particles become large enough to be filtered out or heavy enough to settle so they can be vacuumed out.

Although aluminum sulfate (alum) was the most common flocculant used in the past, today cellulose fiber or poly aluminum chloride are more common. The products are added directly to the filter bed and form a layer on top of or between the grains of sand media.

Clarifiers are biodegradable organic polymers usually made up of the natural polymer chitin often extracted from sea organisms. Positively charged repeating polymer links attract negatively charged colloidal particles. The electric charge is neutralized, and the polymer coils up into a large particle, which can be filtered.

Infrequent vacuuming of debris from the pool can contribute to cloudy water conditions. Make sure the pool is routinely being vacuumed on a daily basis, first thing in the morning, or after a period of quiescence of at least 2 hours, to allow debris which is heavier than water to settle on the bottom of the pool. Check that portable or robotic pool vacuum filters are being disinfected and cleaned properly.

And finally, make sure the pool does not have any circulation dead spots. Perform a dye test of pool circulation patterns to make sure all inlets are operating properly. Note the inlet pattern, any inlets that don’t work, inlets where the water stream is weak, inlets pointed in the wrong direction, or inlets in need of adjustment. Look for circulation eddies or weak spots where water does not change color and record. If filtered, heated, chemically treated water is not being uniformly distributed to all areas of the pool, it is likely algae will become established in the pool, and other water quality problems will develop.

Pool Problems

Pool Problems is an on-going column. Does your pool have a persistent problem? Submit your pool problem and/or pool operations question to sjgrosse@execpc.com. The purpose of this column is to help you, our readers, operate safe, healthful facilities.

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